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Modern Chemistry CHAPTER 7 HOMEWORK 7-2 . CHAPTER 7 HOMEWORK 7-2 VOCABULARY 1. d 2. c 3. e 4. a 5. g 6. h 7. b 8.. Modern Chemistry CHAPTER 7 HOMEWORK 7-1 (pp. 203-206) . 5 (NO. 2. h. a. . MODERN CHEMISTRY SECTION 7-1 REVIEW. 53.. Modern Chemistry CHAPTER 8 HOMEWORK 8-5 VOCABULARY AND SKILL BUILDER Answer each question. 1. What is another name ...

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CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar 2. For each type of investigation, select the most appropriate branch of chemistry from the following

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom. (c) the charge on ...

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Chem 334 - Fall 1998 - Organic Chemistry I Portland State University - Dr. Carl C. Wamser . Chapter 5 Homework Answers. Brown & Foote, pages 175 - 180 : Problems 5.7 - 10, 13 - 30. 1. Write IUPAC names, including designation of stereochemistry where needed, for each of the following compounds: 1a) 1-butyl-6,6-dimethyl-1,4-cyclohexadiene 1b) (2E,7E)-6-ethyl-3,9-dimethyl-2,7-decadiene 1c) (1R,3R ...

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Modern Chemistry • CHAPTER 6 HOMEWORK 6-7 (pp. 179–180) VOCABULARY Write true or false for each statement. 1. A polyatomic ion is a group of covalently bonded atoms. ____ 2. NaCl is a molecular compound. ____ 3. Lattice energy is released when the ions in a crystal lattice are separated from each other. ____ 4.

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Modern Chemistry • CHAPTER 7 HOMEWORK 7-5 (pp. 216–218) VOCABULARY Complete each sentence. 1. Oxidation numbers are assigned to the atoms composing a compound or ion in order to _____. 2. The atoms in a pure element have an oxidation number of _____. 3. The more electronegative element in a binary molecular compound is assigned the number that _____. 4. The algebraic sum of the ...

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Circle the letter of the best answer. 1. Which statement is true? a. In scientific experiments, accuracy and precision mean the same thing. b. Measured values that are precise are usually close to the accepted value. c. Measured values that are accurate are close to the accepted value. d. The term precision can be defined as "highly detailed ...

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Chemistry Homework Answer Key #9. Chapter 12 - Electrochemistry. 19. The galvanic cell given is: Ni(s) | Ni²⁺(aq) || Ag⁺(aq) | Ag(s) E° = 1.03 V oxidation: Ni(s) → Ni²⁺(aq) + 2 e⁻ reduction: Ag⁺(aq) + e⁻ → Ag(s) Net reaction: Ni(s) + 2 Ag⁺(aq) → Ni²⁺(aq) + 2 Ag(s) In this redox reaction, n = 2. 23. In this problem, solid indium is observed to plate out on one of the ...

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CHAPTER 7 • HOMEWORK 7-1 VOCABULARY 1. false 2. true 3. false 4. false 5. true SKILL BUILDER AgCl—In the compound, there is one atom of silver for each atom of chlorine. Ba(OH)₂—The subscript 2 refers to everything inside the parentheses, so there are two hydroxide ions for

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Modern Chemistry • CHAPTER 10 HOMEWORK 10-4 (pp. 311–312) VOCABULARY Select the term in Column B that matches the expression in Column A. Column A Column B 1. a. 1 N/m² 2. millim B of mercury ____ b. mm Hg 3. Pascal ____ c. force/area 4. 1 atmosphere ____ d. 760 mm Hg 5. STP ____ e. 1 atm, 0°C SKILL BUILDER Fill in the chart by supplying the missing ...

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