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Modern Chemistry 35 Chapter Test Name Class Date Chapter Test B, continued Write the orbital notation for the following elements in the space provided. 37. lithium, atomic number 3 38. carbon, atomic number 6 39. neon, atomic number 10 PART VI Write the answers to the questions on the line to the left, and show your work in the space provided.

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The research chemist who is developing a new cancer ... You can test your readiness to proceed by answering the Review Questions at the end of the chapter. This might also be a good time to read the Chapter Objectives, which precede the Review Questions. 9.1 A Typical Problem 1 To avoid potential confusion, the states—(s), (l g), and (aq)—are not mentioned in equations describing ...

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Modern Chemistry 2 Chapter Test Name Class Date Chapter Test A, continued ____ 6. To find a safer chemical for industrial purposes, a scientist is working with chemicals that do not cause depletion of ozone in the atmosphere. This is an example of a. applied research. b. basic research. c. technological development. d. None of the above ____ 7.

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Modern Chemistry 141 Chapter Test Name Class Date Chapter Test B, continued 18. When a strong acid is titrated with a weak base, the pH of the solution at the equivalence point is than 7. 19. A is a highly purified solid used to check the concentration of a standard solution. 20. A 1 M solution of NaOH will have a pH that is

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Modern Chemistry 136 Chapter Test Name Class Date Chapter Test A, continued ____ 14. A pH meter measures the pH of a solution by measuring the a. amount of charge on the hydronium ion. b. color of the solution. c. concentration of the solution. d. voltage difference between two electrodes in the meter's probe. ____ 15. When you conduct an ...

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